

HELOGYNIC ACID, A HOMODITERPENE, AND OTHER CONSTITUENTS FROM *HELOGYNE APALOIDEA*

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Key Word Index—*Helogyne apaloidea*; Compositae; sesquiterpene lactones; germacradienolides; guaianolides; diterpenes; geranyl nerol derivatives; homoditerpene.

Abstract—The aerial parts of *Helogyne apaloidea* afforded in addition to 18 known sesquiterpene lactones four new germacradien-12,6-olides and 10 guaianolides as well as two geranyl nerol derivatives, one being a homoditerpene. The structures were elucidated by high field NMR techniques. The chemistry of this species is similar to that of *H. hutchisonii*.

INTRODUCTION

The South American genus *Helogyne* is placed in the subtribe Alomiinae [1]. The chemistry of one species showed that most likely this genus is a primitive element in the subtribe [2]. We now have studied a further species, from Peru, *H. apaloidea* Nutt.; the results are presented in this paper.

RESULTS AND DISCUSSION

Careful separation of the extract of the aerial parts afforded lupeyl acetate, 6-methoxy- and 6-hydroxy-3 α -angeloyloxytremetone [3], the germacranolides eupatolide [4], 1 [5], 2 [6], 7 [7], 8 [8], 9 [9] and 10 [10], the guaianolides 11 [11], 12 [12], 13 [13], 23 [12], 24 [14], 25 [15], 26 [16], 27 [17] and 28 [16], the heliangolide eucannabinolide [5] and its 20-desoxy derivative [18] as well as four new germacranolides (3-6), ten guaianolides (14-22 and 29), the geranyl nerol derivative 30 and the corresponding homoditerpene 31, named helogynic acid.

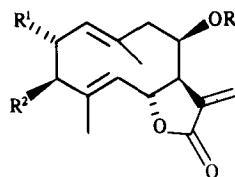
The structures of 3-6 could easily be deduced from their ^1H NMR spectra (Table 1) which were close to those of 1 and 2. The nature of the different ester residues followed from the typical NMR signals and from the mass spectra. Similarly the ^1H NMR spectra of 14-20 (Table 2) were close to that of 12 and the structure of the 8 β -acyloxy groups could also be determined by the NMR data. Spin decoupling allowed the assignment of all signals. The somewhat different ^1H NMR spectrum of 29 was close to that of the corresponding 20-desacetyl derivative [14]. The main difference was the expected downfield shift of the H-5' signals and the presence of an acetoxy methyl singlet.

The acids 30 and 31 were purified as their methyl esters. The mass spectrum of the latter showed the highest ion at m/z 404 corresponding to $\text{C}_{24}\text{H}_{36}\text{O}_5$. The next ion, m/z 362 ($\text{C}_{22}\text{H}_{34}\text{O}_4$), obviously was formed by elimination of acetic acid. Accordingly, the natural product must be the acetate of a C_{21} -trihydroxy acid. This was established by the ^{13}C NMR spectrum of the methyl ester of 31 (see Experimental). Inspection of the ^1H NMR spectrum (see

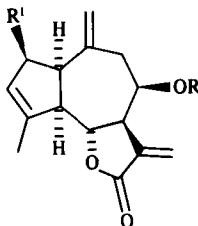
Experimental) further showed that a geranyl nerol derivative with an additional methyl group must be present where only one methyl was not oxygenated. Signals in deuteriobenzene at δ 4.13 (d), 3.95 (s), 4.67 (s) (each 2 H) required three oxygen functions bearing methylene groups. Furthermore, a pair of olefinic methylene protons, an olefinic methyl signal and a methoxy singlet indicated that the necessary equivalents of methyl groups of a geranyl nerol derivative were present. However, an additional methyl doublet at δ 1.14 showed that a homoditerpene was present. Spin decoupling allowed us to assign nearly all signals while the chemical shifts gave some indications about the relative position of the oxygen functions. Final proof was achieved by NOE difference spectroscopy. Thus, a clear effect between H-18 and H-9 indicated an E-configuration of the corresponding double bond. Similar the NOE between H-20 and H-2 established the neryl configuration, while the effect between H-19 and H-5 established the configuration of the remaining trisubstituted double bond. Furthermore the relative position of the olefinic methyl group could be established by the presence of an allylic coupling between H-10 and H-12. The signal of the latter was assigned by spin decoupling starting with H-14 which showed an allylic coupling with H-16'. This also showed that the secondary methyl group was at C-14.

The ^1H NMR spectrum of the methyl ester of 30 (see Experimental) was in part close to that of the methyl ester of 31. However, the changed situation at C-13-C-16 was obvious. The corresponding signals now indicated that the secondary methyl was missing and that the olefinic methylene group was replaced by a trisubstituted double bond as followed from a low field narrowly splitted triplet at δ 5.82 and an additional olefinic methyl group (δ 1.93 d). The configuration of the Δ^{14} bond followed from the chemical shift of H-14.

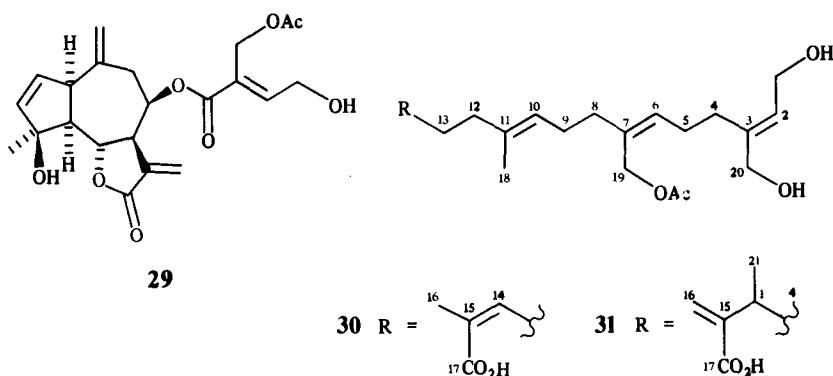
The chemistry of the second *Helogyne* species again shows that most likely this genus is less advanced than the other genera placed in the subtribe Alomiinae where 8 β -substituted germacranolides and guaianolides, which are typical for other less advanced genera in the tribe, are



	1	2	3	4	5	6	7	8	9	10
R	Ac	Tigl	Ang	Prop	<i>i</i> Bu	I	K	B	Ang	C
R¹	H	H	H	H	H	H	H	H	OH	H
R²	H	H	H	H	H	H	H	H	H	OH



	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28
R	H	A	B	C	Ang	D	E	F	G	I	C	D	K	H	Ang	Tigl	A	B
R¹	H	H	H	H	H	H	H	H	H	OH								



A = $\text{CO}(\text{Me})=\text{CHCH}_2\text{OH}$ (E), **B** = $\text{COC}(\text{CH}_2\text{OH})=\text{CHCH}_2\text{OH}$ (E), **C** = $\text{COC}(\text{CH}_2\text{OH})=\text{CHCH}_2\text{OAc}$ (E)
D = $\text{COC}(\text{CH}_2\text{OAc})=\text{CHCH}_2\text{OH}$ (E), **E** = $\text{COC}(\text{CH}_2\text{OAc})=\text{CH}(\text{CH}_2\text{OAc})$ (E), **F** = $\text{COC}(\text{CH}_2\text{OAng})=\text{CHCH}_2\text{OH}$
G = $\text{COC}(\text{CH}_2\text{OTigl})=\text{CHCH}_2\text{OH}$ (E), **I** = β -furoyl, **K** = α -hydroxy-2,5-H-furoyl

Table 1. ^1H NMR spectral data of compounds 3–6 (CDCl₃, 400 MHz, δ -values)

H	3*	4†	5‡	6§	Multiplicity
1	4.87	4.88	4.89	4.91	br dd
5	4.78	4.77	4.78	4.83	br d
6	5.14	5.12	5.13	5.19	dd
7	2.92	2.90	2.91	2.98	br dt
8	5.79	5.74	5.73	5.94	br d
9	2.87	2.82	2.81	2.91	br dd
9'	2.35	2.34	2.34	2.38	br d
13	6.30	6.30	6.30	6.32	d
13'	5.61	5.59	5.58	5.68	d
14	1.48	1.50	1.51	1.50	br s
15	1.73	1.76	1.77	1.81	d

*OAng: 6.10 *qq*, 1.97 *dq*, 1.85 *dq* (*J* [Hz]: 3,4 = 7; 3,5 = 4,5 = 1.5);†OProp: 2.33 *q*, 1.14 *t* (*J* = 7 Hz).‡O*i*Bu: 2.54 *qq*, 1.16 *d*, 1.15 *d* (*J* = 7 Hz).§furoyl: 7.96 *dd* (*J* = 1,1.5 Hz), 7.45 *t* (*J* = 1.5 Hz), 6.68 *br d* (*J* = 1.5 Hz).*J*[Hz]: 1,2 = 11, 1,2' = 4; 5,6 = 10; 5,15 = 1.5; 6,7 = 9; 7,13 = 3.5; 7,13' = 3; 8,9 = 5; 9,9' = 14.

missing. Most genera contain diterpenes. In this connection the isolation of 30 and 31 may be of interest as this could be an indication that the enzymes for the typical cyclization reaction of the alicyclic diterpenes are not present in the more primitive elements of the tribe Eupatoreiae.

EXPERIMENTAL

Air-dried aerial parts (720 g, collected in November 1986 in Peru, voucher M. O. Dillon 4792, deposited in the Herbarium of the Field Museum of Natural History, Chicago, U.S.A.) were extracted and worked-up as reported previously [19]. CC fractions were taken as follows: 1: petrol and Et₂O-petrol (1:9), 2: Et₂O-petrol (1:1), 3: Et₂O, 4: Et₂O-MeOH (9:1) and 5: Et₂O-MeOH (1:1) which were further sepd by TLC silica gel, solvents T1 = Et₂O, T2 = Et₂O-petrol (3:1), T3 = Et₂O-petrol-MeOH (18:6:1) and T4 = Et₂O-petrol-MeOH (36:12:1) and by HPLC RP 18, *ca* 100 bar, solvents: H1 = MeOH-H₂O (4:1), H2 = MeOH-H₂O (7:3) and H3 = MeOH-H₂O (3:2). TLC of fraction 1 (petrol) gave no definite compounds and fraction 2 afforded 100 mg lupeyl acetate. Fraction 3 was sepd by medium pressure chromatography (MPC) using silica gel θ 30–60 and Et₂O-petrol mixts as eluents affording two fractions (3/1 and 3/2). HPLC of 3/1 (H1) gave 4 mg 1, 1 mg 6 (*R*, 3.3 min), 1 mg 4 (*R*, 3.4 min), 1 mg 5 (*R*, 4.3 min), 1 mg 20 (*R*, 4.5 min), 8 mg 2, 80 mg 3 (*R*, 5.6 min), 4 mg 15 (*R*, 7.3 min), 3 mg 6-methoxy-3 α -angeloyloxytremetone and 3 mg of the 6-hydroxy derivative. HPLC of 3/2 (H1) gave a mixt. (3/2/1) (*R*, 1.8 min), 4 mg 17 (*R*, 3.5 min), 20 mg 11 (*R*, 6.0 min), 10 mg 12 (*R*, 8.0 min), a mixt. (3/2/2) (*R*, 12.3 min), a mixt. (3/2/3) (*R*, 13.5 min) and 30 mg 18 (*R*, 15.1 min). TLC of 3/2/1 (T2) gave 2 mg 25 and 3 mg 26. TLC of 3/2/2 (T2, two developments) gave 2 mg 14 and 5 mg 16. TLC of 3/2/3 (T2) gave 10 mg 18 and 10 mg 19 (*R*, 0.35). CC fraction 4 was sepd by MPC (Et₂O, Et₂O-MeOH (3:2) into 8 fractions (4/1–4/8). HPLC of 4/1 (H2) afforded 2 mg 7, 3 mg 9, a mixt. (4/1/1), a mixt. (4/1/2), 20 mg 19 (*R*, 9.0 min) and 60 mg 18 (*R*, 9.8 min). TLC of 4/1/1 (T1) gave 2 mg 23, 3 mg eupatolide, 10 mg 11 and 60 mg 16 (*R*, 0.50). TLC of 4/1/2 (T2, two developments) afforded 20 mg 12, 20 mg 14 (*R*,

0.45) and 10 mg 16. Fraction 4/2 contained 1 g 16 and fraction 4/3 500 mg coumaric acid. HPLC of fraction 4/4 (H2) gave 5 mg 27 (*R*, 1.7 min), 2 mg 20-desoxyeucannabinolide (*R*, 2.1 min) and 3 mg 9 (*R*, 3.3 min). HPLC of 4/5 (H2) gave 30 mg 22 (*R*, 1.3 min), a mixt. (4/5/1) (*R*, 1.5 min), 10 mg 8 (*R*, 2.2 min), 30 mg 13 (*R*, 2.7 min), 15 mg 30 (*R*, 4.9 min) and 20 mg 31 (*R*, 9.8 min). Fraction 4/5/1 gave by TLC (T4) 5 mg 27 and 3 mg 21 (*R*, 0.28). HPLC of fraction 4/6 (H3) gave 20 mg 10, a mixt. (4/6/1) (*R*, 2.6 min), 20 mg 22 (*R*, 3.2 min), 100 mg 8 and 100 mg 13. TLC of 4/6/1 (T3) gave 5 mg 24 and 8 mg 29 (*R*, 0.30). HPLC of 4/7 (H3) gave 10 mg 10 and 60 mg eucannabinolide. Fraction 4/8 gave 1 g eucannabinolide. From fraction CC-5 5% was sepd by HPLC (H3) affording 50 mg eucannabinolide, 25 mg 28, 25 mg 8 and 20 mg 13. Known compounds were identified by comparing the 400 MHz ^1H NMR spectra with those of authentic material. The data of the new lactones are given in Table 3.

19-Acetoxy-20-hydroxygeranylnerol-17-acid (30). Colourless oil which was purified as its Me ester, colourless oil, IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 3600 (OH), 1745 (OAc), 1720 (C=CCO₂R); MS *m/z* (rel. int.): 390.241 [M-H₂O]⁺ (0.1) (calc. for C₂₃H₃₄O₅: 390.241), 348 [M-HOAc]⁺ (1.2), 330 [348-H₂O]⁺ (5), 93 (100); ^1H NMR (C₆D₆): δ 4.12 (*d*, H-1), 5.70 (*br t*, H-2), 2.20 (*m*, H-4, H-5, H-8, H-9), 5.31 (*br t*, H-6), 5.29 (*br t*, H-10), 2.14 (*br t*, H-12), 2.80 (*br q*, H-13), 5.82 (*tq*, H-14), 1.93 (*d*, H-16), 1.66 (*s*, H-18), 4.66 (*s*, H-19), 4.12 (*br d*, H-20), 1.74 (*s*, OAc) 3.44 (*s*, OMe) (*J* [Hz]: 1,2 = 5,6 = 9,10 = 12,13 = 13,14 = 7; 14,16 \pm 1.5).

Healogenic acid (31). Colourless oil which was purified as its Me ester (CH₂N₂-Et₂O, 3 min, 20°), colourless oil; IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 3600 (OH), 1720 (CO₂R); MS *m/z* (rel. int.): 404 [M-H₂O]⁺ (0.05), 362.246 [M-HOAc]⁺ (3) (calc. for C₂₂H₃₄O₄: 362.246), 344 [362-H₂O]⁺ (5), 326 [344-H₂O]⁺ (4), 313 [344-OMe]⁺ (5), 261 (8), 213 (15), 135 (76), 93 (92), 81 (100); ^1H NMR (C₆D₆): δ 4.13 (*br d*, H-1), 5.71 (*br t*, H-2), 2.10 (*t*, H-4), 2.19 *q* (H-5), 5.36 (*br t*, H-6), 2.22 (*br s*, H-8, H-9), 5.27 (*br t*, H-10), 2.08 and 2.04 (*m* H-12), 1.80 and 1.60 (*dd* H-13), 2.84 (*tq*, H-14), 5.36 (*t*, H-16), 6.24 (*d*, H-16'), 1.65 (*d*, H-18), 4.67 (*s*, H-19), 3.95 (*br s*, H-20), 1.14 (*d*, H-21), 1.74 (*s*, OAc) 344 (*s*, OMe) (*J* [Hz]: 1,2 = 3,4 = 5,6 = 9,10 = 12,13 = 13,14 = 13',14 = 13,21 = 7; 10,18 = 14,16' = 16,16' = 1; 13,13' = 14); ^{13}C NMR (C₆D₆, C-1-C-21): δ 58.5 *t*, 130.2 *d*, 134.5 *s*, 34.5 *t*, 28.2 *t*, 124.1 *d*, 146.2 *s*, 37.4 *t*,

Table 2. ^1H NMR spectral data of compounds **14–22** and **29** (CDCl_3 , 400 MHz, δ -values)

H	14*	15†	16‡	17§	18 	19¶	20**	21††	22‡‡	29§§	Multiplicity
1	3.17	3.16	3.14	3.17	3.15	3.15	3.19	3.17 <i>dd</i>	3.14 <i>dd</i>	3.37 <i>br d</i>	<i>ddd</i>
2	2.42	2.52	2.50	2.52	2.52	2.51	2.50	4.74 <i>br s</i>	4.70 <i>br s</i>	5.76 <i>dd</i>	<i>m</i>
2'	2.33	2.43	2.41	2.43	2.40	2.41	2.41				<i>br d</i>
3	5.57	5.56	5.56	5.58	5.56	5.56	5.57	5.77	5.72	5.95	<i>br s</i>
5	2.74	2.85	2.86	2.86	2.84	2.85	2.87	2.68	2.65	2.42	<i>br t</i>
6	4.49	4.51	4.52	4.51	4.50	4.51	4.51	4.66	4.65	4.97	<i>dd</i>
7	3.20	3.19	3.17	3.20	3.18	3.18	3.21	3.20	3.19	3.24	<i>dddd</i>
8	5.65	5.64	5.65	5.67	5.66	5.73	5.67	5.59	5.58	5.70	<i>ddd</i>
9	2.48 <i>d</i>	2.61	2.60	2.61	2.58	2.59	2.66	2.83	2.82	2.82	<i>dd</i>
9'		2.56	2.52	2.55	2.53	2.52	2.60	2.78	2.78	2.38	<i>dd</i>
13	6.29	6.30	6.26	6.29	6.24	6.24	6.29	6.30	6.26	6.32	<i>d</i>
13'	5.54	5.56	5.55	5.56	5.54	5.54	5.60	5.53	5.52	5.64	<i>d</i>
14	5.03	5.00	5.00	5.03	4.99	5.00	5.04	5.14	5.12	4.96	<i>br s</i>
14'	4.89	4.88	4.87	4.89	4.85	4.87	4.91	5.12	5.08	4.94	<i>br s</i>
15	1.79	1.88	1.87	1.89	1.87	1.87	1.90	2.02	1.98	1.57	<i>br s</i>

OCOR: * 6.67 *t* ($J = 5.5$), 4.82 *d* ($J = 5.5$), 4.33 *br s*, 2.01 *s*.† OAng: 6.07 *qq*, 1.93 *dq*, 1.80 *dq*.‡ OCOR: 6.99 *t* ($J = 5.5$) 4.45 *br t*, 4.88 and 4.75 *d* ($J = 12$), 1.98 *s*.§ OCOR: 6.89 *t* ($J = 5.5$) 4.88 *d* ($J = 5.5$), 4.80 and 4.75 *d* ($J = 12$), 2.00 and 2.11 *s*.¶ OCOR: 6.98 *t* ($J = 5.5$) 4.45 *d* ($J = 5.5$), 4.93 and 4.89 *d* ($J = 12$), 6.08 *qq* ($J = 7.1, 5$), 1.93 *dq* ($J = 7.1, 5$), 1.82 *dq* ($J = 1.5, 1.5$).|| OCOR: 7.01 *t* ($J = 5.5$), 4.50 *d* ($J = 5.5$), 4.94 and 4.89 *d* ($J = 12$), 6.77 *qq* ($J = 7.1$), 1.78 *dq* ($J = 7.1$), 1.77 *br s*.** 7.92 *br s*, 7.42 *t* ($J = 1.5$).†† OCOR: 6.68 *t* ($J = 6$), 4.81 *dd* ($J = 6.2$), 4.32 *br s*, 2.12 *s*.‡‡ 6.97 *t* ($J = 5.5$), 4.45 and 4.40 *dd* ($J = 15.5$), 4.87 and 4.74 *d* ($J = 12$), 1.99 *s*.§§ OCOR: 7.02 *t* ($J = 5.5$), 4.48 and 4.43 *dd* ($J = 16.5$), 4.90 and 4.76 *d* ($J = 12$), 1.29 *s*.J[Hz]: 1,2 = 1.5~10; 1,2' = 5; 2,2' = 16; 5,6 = 10; 6,7 = 8.5; 7,8 = 3; 7,13 = 3.5; 7,13' = 3.8; 9 = 4.5; 8,9 = 5; 9,9' = 14; compounds **21–22**: 1,2 = 6; compound **29**: 1,2 = 2.5; 1,3 = 1.5; 2,3 = 5.5; 8,9 = 8.9' = 3.

Table 3. Physical data of compounds 3-6, 14-22 and 29.

	OH	IR ν	CO_2R	solvent	MS m/z (rel. int.) [M] ⁺	Calc. for	main fragments	$[\alpha]_{\text{D}}^{24^{\circ}}$ (CHCl ₃ ; c)
3*	—	1770	1715	CCl ₄	330.183 (1)	C ₂₀ H ₂₆ O ₄	230 (26), 215 (8), 83 (100), 55 (92)	+68
4	—	1780	1745	CCl ₄	304.167 (1)	C ₁₈ H ₂₄ O ₄	230 (28), 215 (12), 57 (100)	0.91
5	—	1775	1740	CCl ₄	318.183 (1)	C ₁₉ H ₂₆ O ₄	230 (51), 215 (19, 71 (100)	
6	—	1775	1725	CCl ₄	342.147 (0.5)	C ₂₀ H ₂₂ O ₅	230 (26), 215 (18), 95 (100)	
14	3480	1770	1745,	CCl ₄	402.168 (4.5)	C ₂₂ H ₂₆ O ₇	228 (100), 213 (13), 115 (69), 97 (57)	
15	—	1780	1720	CCl ₄	328.167 (5)	C ₂₀ H ₂₄ O ₄	228 (27), 213 (6), 83 (100), 55 (55)	1.86
16	3600	1770	1740,	CCl ₄	402.168 (2.5)	C ₂₂ H ₂₆ O ₇	228 (100), 213 (26), 157 (43), 115 (37), 97 (59)	+43
17	—	1770	1725,	CCl ₄	444.178 (2)	C ₂₄ H ₂₈ O ₈	384 (2.5), 228 (53), 157 (29), 115 (52), 55 (100)	+40
18	3610	1775	1720	CCl ₄	442.199 (2)	C ₂₅ H ₃₀ O ₇	228 (36), 83 (100), 55 (45)	+40
19	3620	1775	1720	CCl ₄	442.199 (1.5)	C ₂₅ H ₃₀ O ₇	228 (27), 83 (100), (55) (49)	
20	—	1775	1725	CCl ₄	340.131 (6)	C ₂₀ H ₂₀ O ₅	228 (22), 95 (100)	
21	3620	1760	1740,	CHCl ₃	418.163 (0.5)	C ₂₂ H ₂₆ O ₈	358 (2), 244 (27), 226 (37), 115 (100), 97 (89)	+36
22	3610	1760	1720	CHCl ₃	418.163 (1)	C ₂₂ H ₂₆ O ₈	358 (2), 244 (32), 226 (35), 115 (80), 97 (100)	-26
29				CHCl ₃	403.139 (3.5)	C ₂₁ H ₂₃ O ₈	343 (5), 244 (12), 226 (26), 115 (81), 97 (100)	4.38
				[M - Me] ⁺			-42	0.79

* Mp 89°, all other colourless oils.

27.1 *t*, 125.5 *d*, 141.1 *s*, 35.4 *t*, 26.7 *t*, 34.5 *d*, 135.2 *s*, 122.5 *t*, 167.2 *s*, 15.8 *q*, 66.2 *t*, 61.4 *t*, 19.6 *q*; OMe: 51.1 *q*; OAc: 20.3 *q*, 170.3 *s* (a few triplets may be interchangeable).

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SPHAEROPYRANE, A DITERPENE FROM THE MARINE RED ALGA *SPHAEROCOCCUS CORONOPIFOLIUS*

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Key Word Index—*Sphaerococcus coronopifolius*; Rhodophyta; diterpenes; sphaeropyrane.

Abstract—A new diterpene, sphaeropyrane has been isolated from the marine red alga *Sphaerococcus coronopifolius*. Its structure has been assigned mainly on the basis of ¹H and ¹³C NMR experiments.

INTRODUCTION

The cosmopolitan bright-red alga *Sphaerococcus coronopifolius*, generally growing on rocks in shallow places, has been the subject of numerous studies for secondary metabolites [1–13]. Until now 18 new diterpenes based on six new rearranged polycyclic skeletons have been isolated and their biogenesis has been hypothesized [9, 10, 11]. This paper reports on the isolation of sphaeropyrane (**1**), a compound which belongs to a new class of diterpenes with a non-rearranged carbon skeleton.

Stereostructure **1** (relative stereochemistry) has been based on spectral data including extensive spin

decoupling experiments and 2D ¹³C–¹H shift correlated NMR spectroscopy via ¹J, ²J and ³J.

RESULTS AND DISCUSSION

A chloroform extract of the freeze-dried *S. coronopifolius*, after repeated chromatographies over silica gel, provided crude **1** which was purified by reversed-phase HPLC to yield **1** (0.003% dry weight), C₂₀H₃₂O (HRMS), mp 136–138°.

Sphaeropyrane was recognized to possess a $\begin{array}{c} | \\ -C-O-C- \\ | \end{array}$ group by IR (ν_{max} 1025, 1070 and 1120 cm^{−1}) and by